## A DIKETOPIPERAZINE DERIVED FROM TRICHLOROLEUCINE FROM THE SPONGE DYSIDEA HERBACEA

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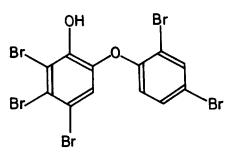
<u>Dysidea</u> <u>herbacea</u> is a Dictyoceratid sponge which is common throughout the Indo-Pacific region. Previous studies<sup>1-3</sup> have shown that the secondary metabolites derived from various collections of this sponge are diverse, ranging from polybromodiphenyl ethers, of which (1) is an example<sup>1</sup>, to compounds derived from amino-acids and containing trichloromethyl-groups. Two examples dysidin (2) and dysidenin (3) have recently been reported<sup>2,3</sup>. We now report a further variant of this sponge, collected near Gladstone, Queensland, Australia, which has yielded a diketopiperazine derivative (4) of an N-methyltrichloroleucine.

Extraction of a small freeze dried sample of <u>D. herbacea</u> with  $CH_2Cl_2$  followed by p.l.c. separation of the extract on silica gel  $(CH_2Cl_2/EtOAc: 4/1)$  yielded (4) m.p. 106-107°,  $[\alpha]_D^{20}$  -144° (c=0.5,  $CHCl_3$ ) in 1% yield. The formula  $C_{14}H_{18}Cl_6N_2O_2$  was established by high resolution mass spectrometry and the presence of two trichloro-methyl groups in the molecule was indicated by the <sup>13</sup>C n.m.r. spectrum which showed two singlets at 105.0 and 104.1 typical of this grouping<sup>2,3</sup>. The remainder of the <sup>13</sup>C n.m.r. consisted of resonances at 165.4 (2) and 159.7 (s) (2 x O=C-N-); 131.6 (s) and 123.7 (d) (-CH=C-); 61.9 (d), 51.5 (d, 2C), 37.7 (t), 33.8 (q), 31.2 (q), 18.4 (q) and 17.6 (q). Compound (4) was therefore monocyclic with two carbonyl groups and one trisubstituted double bond.

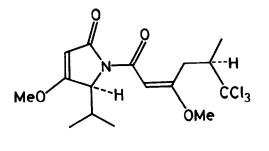
The <sup>1</sup>H n.m.r. spectra of (4) in CDCl<sub>3</sub> and  $C_6D_6$  is shown in Table 1. Extensive decoupling suggested partial structures (5) and (6) and the presence of two -N-Me singlets, together with i.r. and <sup>13</sup>C n.m.r. data, indicated the presence of two -CO-N-Me groups. The diketopiperazine structure (4) best fitted these data and the stereochemistry about the double bond was deduced from the unusually low field position of the C4 proton in the <sup>1</sup>H n.m.r. spectrum of (4) and (7) (see below), due to deshielding by the amide carbonyl group.

Reduction of (4) with zinc-acetic acid gave the dechlorinated analogue (7), m.p. 99-100°  $\left[\alpha\right]_{D}^{20}$  -86° (c=0.5, CHCl<sub>3</sub>), the <sup>1</sup>H n.m.r. spectrum of which (Table 1) was entirely consistent with the proposed structure. The mass spectrum showed a molecular ion at m/e 252 with the major fragment ion at m/e 195 (100%; M<sup>+</sup>-C<sub>4</sub>H<sub>9</sub>). Hydrogenation of (7) gave the single optically active compound (8) m.p. 90-95°  $\left[\alpha\right]_{D}^{20}$  +60° (c=0.6, CHCl<sub>3</sub>).

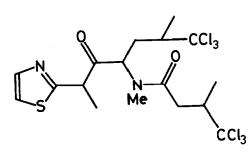
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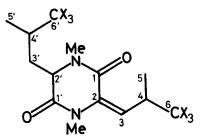


(1)



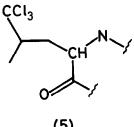
(2)



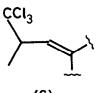


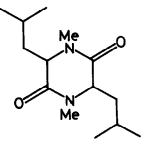
(4) X = CI(7) X = H











(8)

The extreme variability of the secondary metabolites of <u>D. herbacea</u> is curious, particularly in view of the nature of these metabolites. The presence of either polybrominated diphenyl ethers<sup>1</sup> or polychlorinated amino acid derivatives<sup>2,3</sup> has been documented and the biosynthetic origin of both these classes of compounds contrasts markedly with isolates, usually terpenoid, reported from other members of the sponge Family Dysidideae.

A microscopic examination of various forms of <u>D. herbacea</u>, which is by habit a thin encrusting sponge, has revealed the presence of blue-green algae, at times representing half the cellular weight of the sample<sup>4</sup>. Amino acid derived metabolites have been reported from bluegreen algae and the diverse secondary metabolites obtained from separate collections of <u>D. herbacea</u> are most readily rationalised as arising from different blue-green algal symbionts.

## TABLE 1

Chemical shifts and coupling constants of the diketopiperazine (4) and some derivatives

	(4):- CDC1 <sub>3</sub>	(4):- benzene-d <sub>6</sub>	(7):- benzene-d <sub>6</sub>	(8):- CC1 <sub>4</sub>
C2'-H	4.04,1H,dd		3.62,1H,dd,J6,6Hz	2.70,2H,dd,J6,8Hz(+C2-H)
СЗ -Н	5.69,1H,d,J9Hz		5.00,1H,d,J9Hz	
C3'-H_	1.76,1H,m 2.47,1H,m	1.54,1H,m	1.44,2H,m	1.6,4H,m
C3'-H <sub>R</sub>		2.36,1H,m		
С4 -Н	5.18,1H,dq,J6,9Hz	5.50,2H,m	3.96,H,m	1.9,2H,m
.C4'-H	2.70,1H,m	2.64,1H,m	1.70,1H,m	
C5 –H	1.38,6H,d,J6Hz	1.12,3H,d,J6Hz	0.94,3H,d,J6Hz	0.98,6H,d,J6Hz
C5'-H		1.31,3H,d,J6Hz	d,J6Hz 0.69,3H,d,J6Hz	
C6 -H			1.15,3H,d,J6Hz	
C6'-H			0.79,3H,d,J6Hz	1.02,6H,d,J6Hz
N-CH3	3.08,3H,s	2.56,3H,s	2.59,3H,s	2.96,6H,s
N-CH3	3.22,3H,s	2.82,3H,s	2.78,3H,s	

## REFERENCES

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